

Supporting information

Stretchable flexible electrodes based on in situ coating of graphite with PANI

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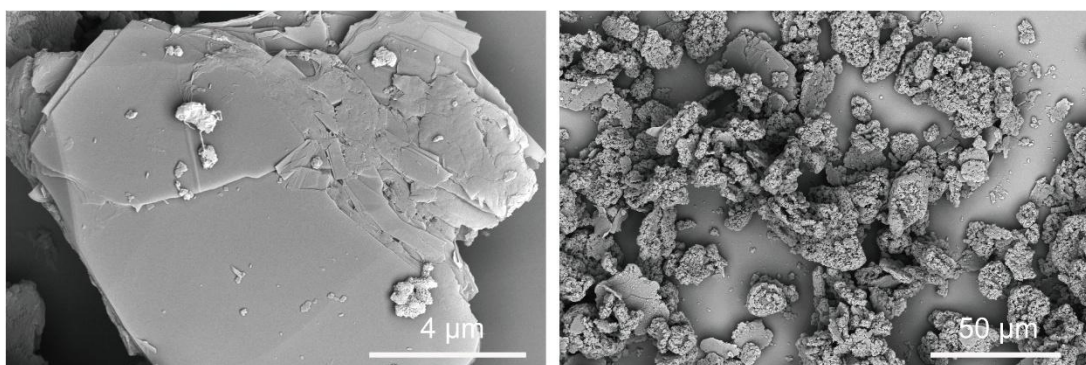


Fig. S1. (Color online) SEM image of the graphite powder surface and PANI@G at low magnification.

For the standard preparation batch, 0.10 g of graphite and 0.50 g of aniline were used, corresponding to a total graphite-and-aniline feed mass of 0.60 g. After polymerization, filtration, washing, and vacuum drying, 0.48 g of dry PANI@G product was recovered. The isolated solid recovery was therefore calculated as:

$$Y_{isolated} = \frac{0.48}{0.10+0.50} \times 100\% = 80\%. \quad (\text{Eq. S1})$$

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The product was collected using a 0.22 μm PTFE membrane. No graphite breakthrough was observed during filtration, and graphite loss through the membrane was therefore considered negligible under the present operating conditions. On this basis, the graphite mass retained in the recovered product was taken as approximately 0.10 g, while the remaining 0.38 g was assigned operationally to the PANI-derived solid fraction. The apparent mass fractions were therefore estimated as approximately 20.8 wt% graphite and 79.2 wt% PANI-derived material:

$$W_{\text{PANI-derived}} = \frac{0.48-0.10}{0.48} \times 100\% = 79.2\%. \quad (\text{Eq. S2})$$

We emphasize that 79.2 wt% is an operational mass-balance estimate rather than an exact molecular-level PANI composition or aniline conversion. The oxidative polymerization process, removal of soluble low-molecular-weight oligomers during washing, and possible contributions from the protonation/deprotonation state of PANI may affect the final dry mass. The unrecovered mass is therefore attributed primarily to soluble low-molecular-weight PANI oligomers and other soluble species removed during washing, rather than to graphite leakage through the filter.



Fig. S2. (Color online) Stretching and torsion of the SEBS substrate.

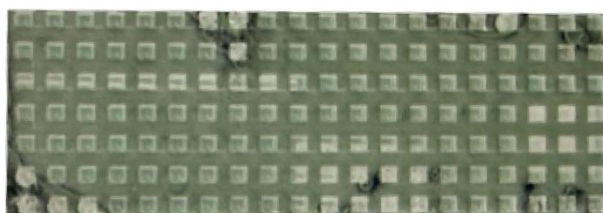


Fig. S3. (Color online) Photograph of laser-patterned grid electrodes.