

Supporting Information for

Synthesis of P-Type PbS Quantum Dot Ink via Inorganic Ligand Exchange in Solution for High-Efficiency and Stable Solar Cells

Napasuda Wichaiyo¹, Yuyao Wei^{1,†}, Chao Ding^{1,†,*}, Guozheng Shi¹, Witoon Yindeesuk², Liang Wang¹, Huān Bì¹, Jiaqi Liu¹, Shuzi Hayase¹, Yusheng Li¹, Yongge Yang¹, and Qing Shen^{1,†}

¹ Graduate School of Engineering Science, The University of Electro-Communications, Tokyo 1828585, Japan

² School of Science, King Mongkut's Institute of Technology Ladkrabang, Bangkok 10520, Thailand

Corresponding authors:

[†]Yuyao Wei Email: wyyelsa@gmail.com

[†]Chao Ding Email: dc1107@scu.edu.cn

[†]Qing Shen Email: qingshen3@yahoo.co.jp

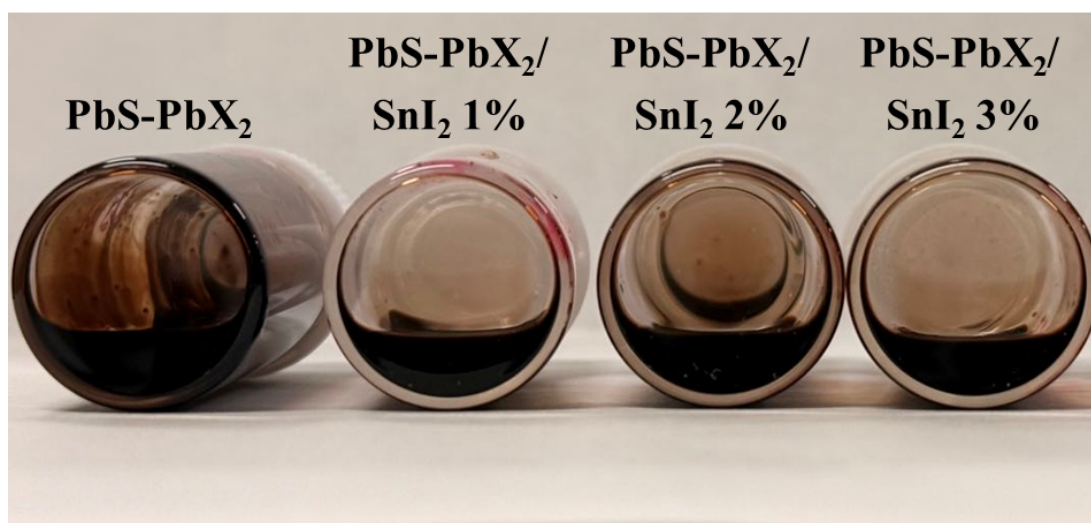


Fig. S1. PbS-PbX₂ ink and PbS-PbX₂/SnI₂ 1%–3% inks (left to right) were dispersed in mixed-amine

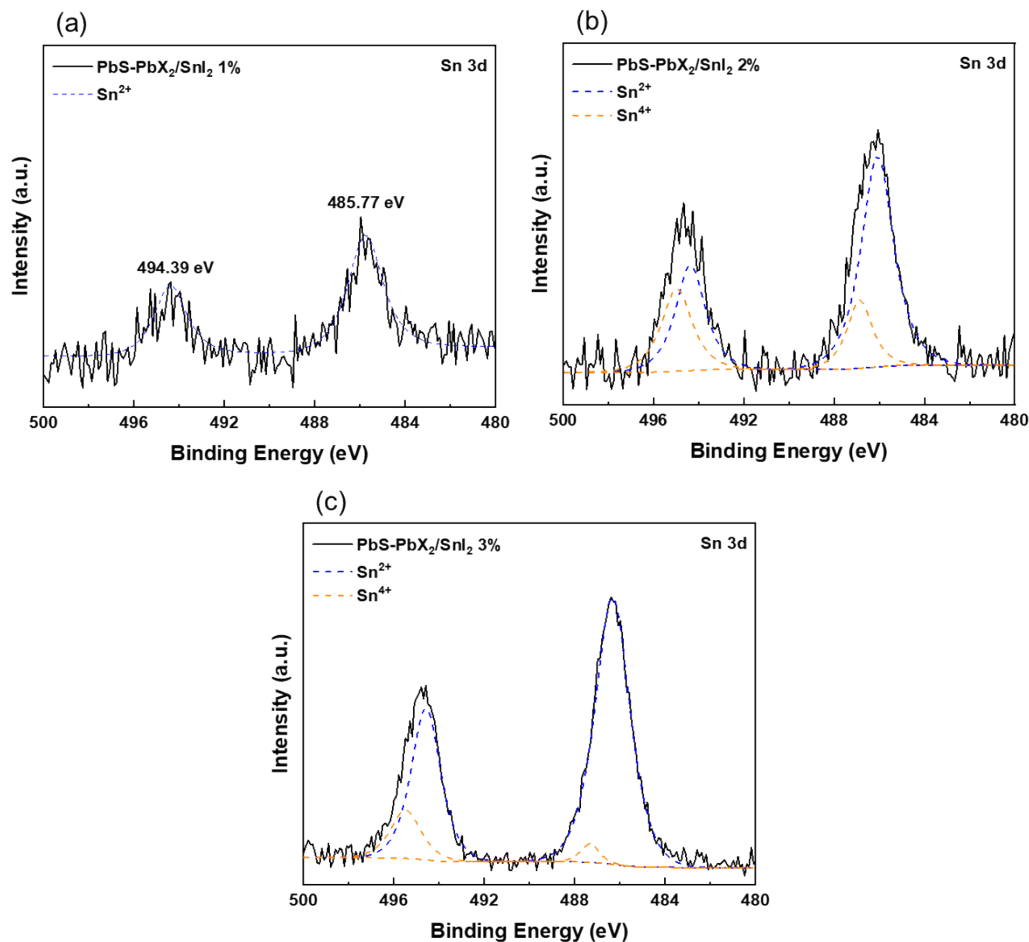


Fig. S2. XPS spectra of Sn 3d region of (a) PbS-PbX₂/SnI₂ 1%, (b) PbS-PbX₂/SnI₂ 2%, and (c) PbS-PbX₂/SnI₂ 3% with two components of Sn²⁺ (blue dash line) and Sn⁴⁺ (orange dash line)

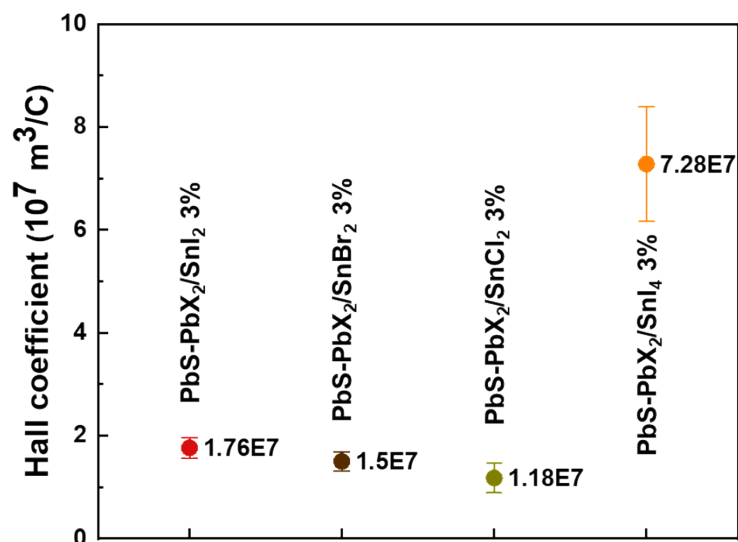
Table S1. The area ratio of Sn²⁺ and Sn⁴⁺ components in PbS-PbX₂/SnI₂ samples (mixing SnI₂ 1%–3%)

Samples	Oxidation state	Centre (eV)	Intensity (eV)	Area (cps*eV)	Ratio* (%)
PbS-PbX ₂ /SnI ₂ 1%	Sn ²⁺	485.8	159.1	306.6	100
		494.4	96.0	169.3	
PbS-PbX ₂ /SnI ₂ 2%	Sn ²⁺	486.1	291.6	525.2	55.28
		494.4	146.3	241.3	
	Sn ⁴⁺	486.9	94.4	132.6	44.72
		495.3	114.1	195.2	
PbS-PbX ₂ /SnI ₂ 3%	Sn ²⁺	486.3	738.7	1451.1	75.42
		494.6	419.9	704.9	
	Sn ⁴⁺	487.3	52.4	52.9	24.58
		495.5	134.8	229.8	

*The area ratio of Sn²⁺ and Sn⁴⁺ component of Sn 3d_{3/2} peak, denoted as X, was determined using the formula:

$$Ratio\ (%) = \left(\frac{X}{Sn^{2+} + Sn^{4+}} \right) \times 100$$

Where X is the area of Sn²⁺ or Sn⁴⁺.



The mixing tin halide composition at 3%

Fig. S3. The hall coefficient of mixing SnX_2 ($\text{X} = \text{I}, \text{Br}, \text{Cl}$) and SnI_4 at 3% (The dots show average values, and the error bars represent the top and lower margins of the 95% confidence interval.)

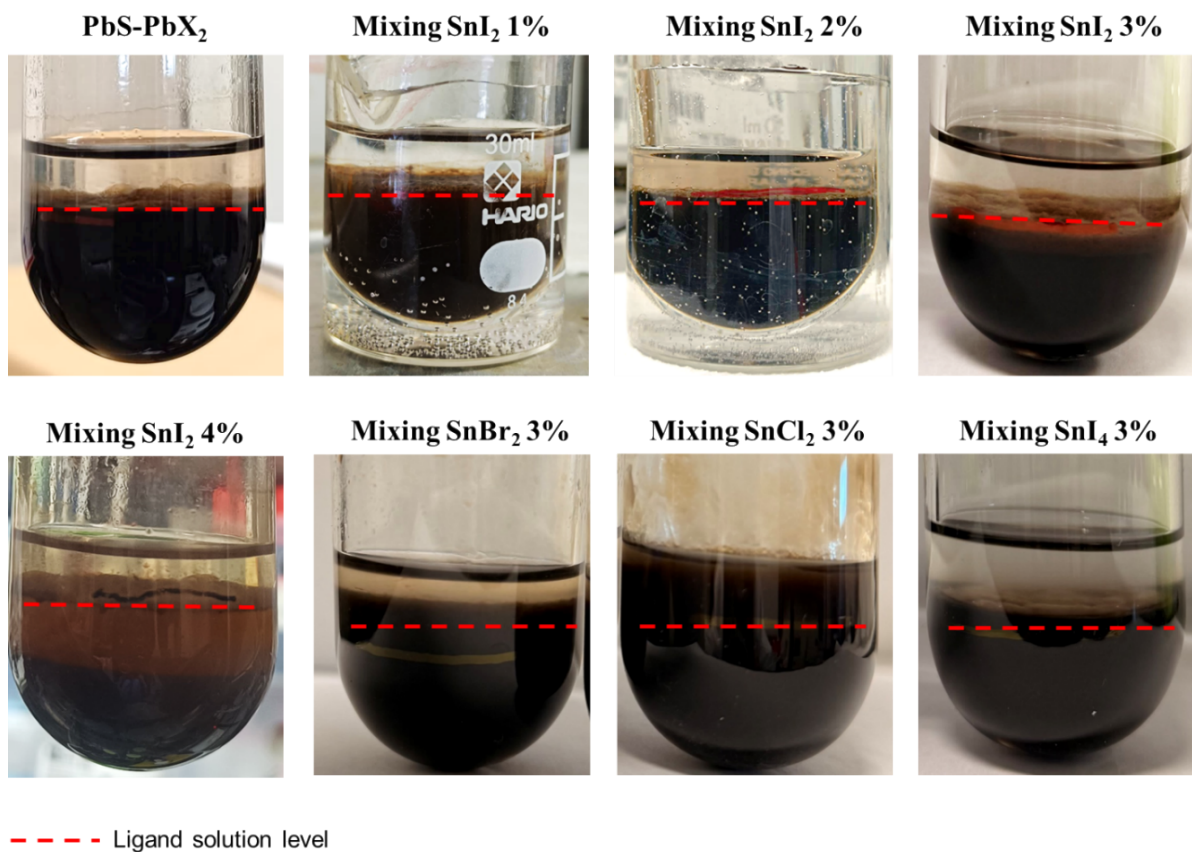


Fig. S4. Photographs of the SPLE technique of PbS CQD in PbX_2 , mixing SnI_2 1%–4%, mixing SnBr_2 3%, mixing SnCl_2 3%, and mixing SnI_4 3% ligand solutions

Table S2. The element ratio-to-S of PbS CQD film without and with mixing SnI₂ 1%–3%

Samples	Pb/S	Sn/S	I/S	Br/S
PbS-PbX ₂	1.963	0	0.686	0.217
PbS-PbX ₂ /SnI ₂ 1%	2.062	0.054	0.714	0.233
PbS-PbX ₂ /SnI ₂ 2%	2.121	0.156	0.720	0.251
PbS-PbX ₂ /SnI ₂ 3%	2.164	0.297	0.763	0.269

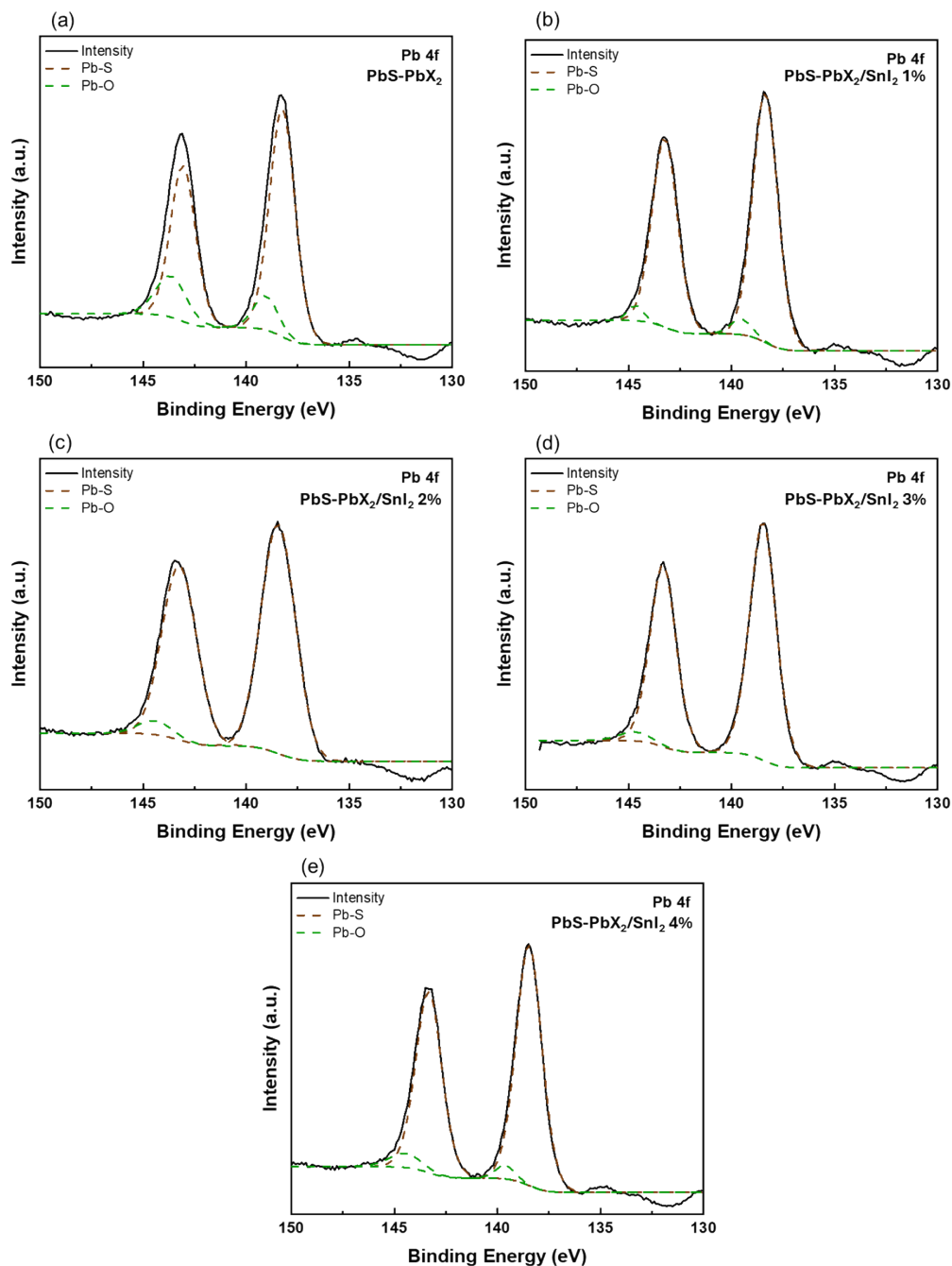


Fig. S5. XPS spectra Pb 4f region of (a) PbS-PbX₂, (b) PbS-PbX₂/SnI₂ 1%, (c) PbS-PbX₂/SnI₂ 2%, (d) PbS-PbX₂/SnI₂ 3%, and (e) PbS-PbX₂/SnI₂ 4%

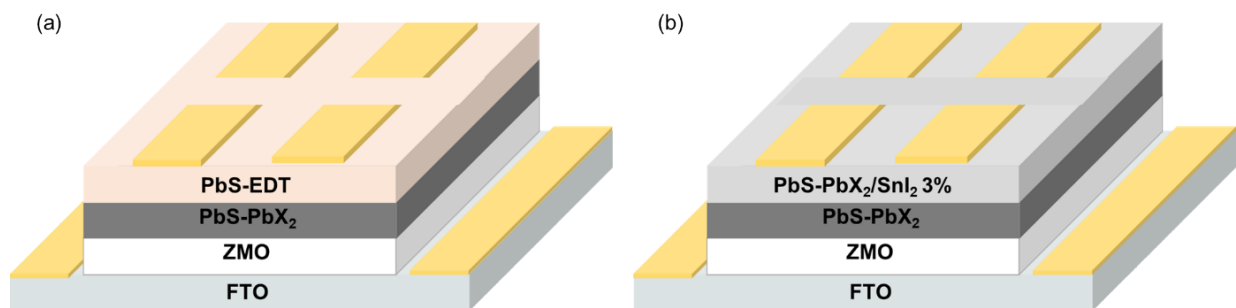


Fig. S6. The schematic structure of the (a) control device (PbS-EDT HTL), (b) target device (p-type PbS ink HTL)

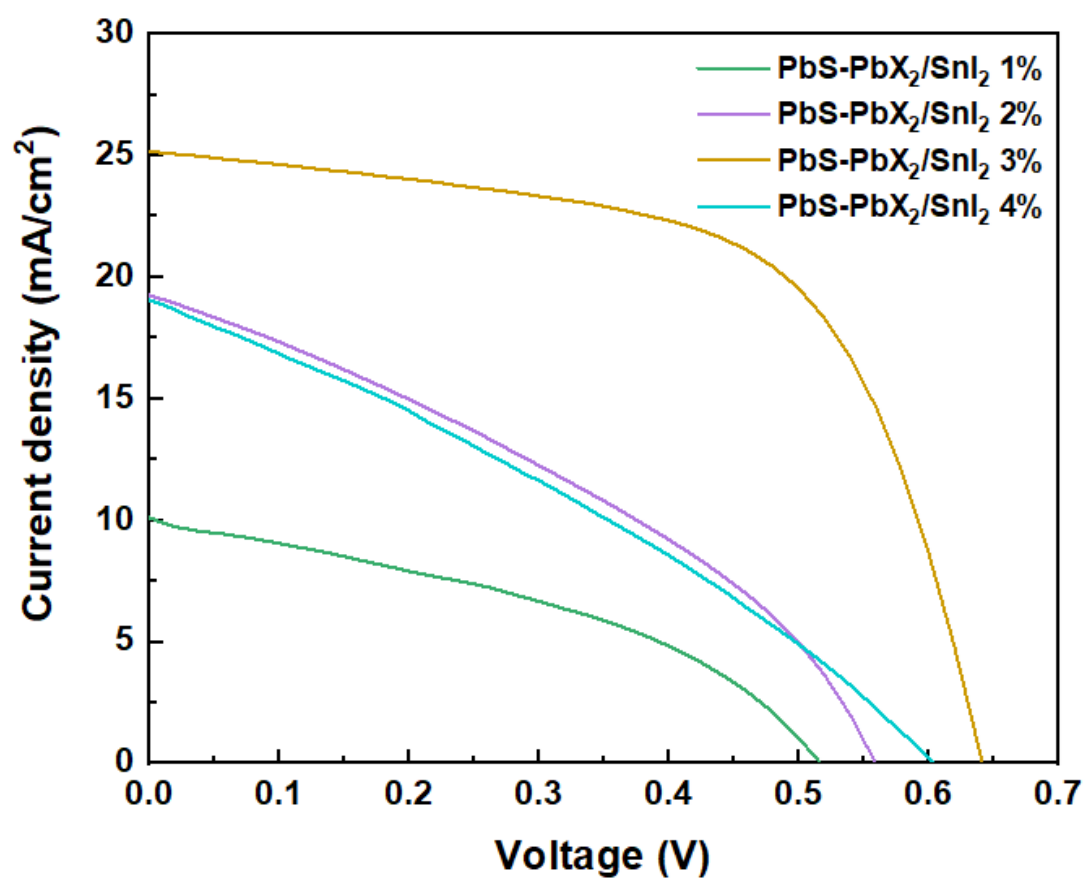


Fig. S7. The comparison J-V characteristics of champion device based on PbS-PbX₂/SnI₂ (1%–4%) HTLs at the fresh state

Table S3. Statistics for device performance of PbS CQD solar cells and champion device

	Device	J_{sc} [mA/cm ²]	V_{oc} [V]	FF	PCE [%]
Fresh state	Control	26.22±1.26	0.63±0.01	0.54±0.01	9.31±0.33
		(29.44)	(0.63)	(0.53)	(10.54)
	Target	25.39±0.49	0.63±0.01	0.57±0.01	9.60±0.27
		(25.15)	(0.64)	(0.61)	(10.37)
1 week storage	Control	26.74±0.55	0.61±0.01	0.57±0.02	9.69±0.14
		(29.58)	(0.63)	(0.52)	(10.42)
	Target	26.79±0.60	0.62±0.01	0.57±0.01	9.95±0.13
		(28.20)	(0.60)	(0.60)	(10.81)
4 weeks storage	Control	27.28±0.85	0.61±0.01	0.51±0.02	8.93±0.36
		(30.18)	(0.61)	(0.50)	(9.83)
	Target	26.91±0.75	0.62±0.01	0.59±0.01	10.33±0.12
		(29.57)	(0.61)	(0.58)	(10.93)
6 weeks storage	Control	28.58±0.24	0.61±0.01	0.45±0.03	7.87±0.63
		(29.11)	(0.61)	(0.55)	9.78
	Target	26.92±0.66	0.61±0.01	0.54±0.01	8.98±0.35
		(28.83)	(0.61)	(0.58)	(10.30)
8 weeks storage	Control	26.69±0.85	0.59±0.01	0.47±0.01	7.37±0.37
		(27.60)	(0.59)	(0.51)	(8.24)
	Target	25.73±0.47	0.60±0.01	0.53±0.01	8.12±0.29
		(27.70)	(0.60)	(0.57)	(9.38)

* Results of J_{sc} , V_{oc} , FF, and PCE for the champion devices are shown in parentheses. The statistics are based on 8 cells for each condition.